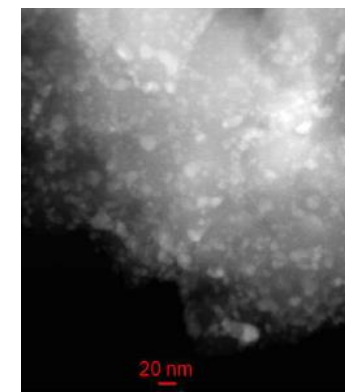
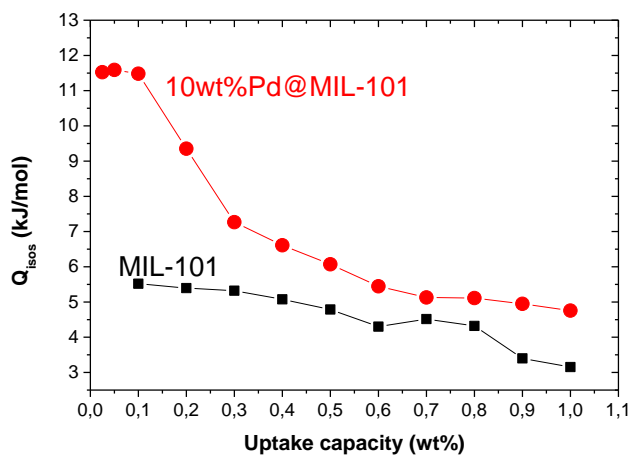
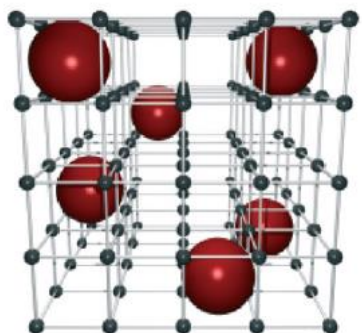
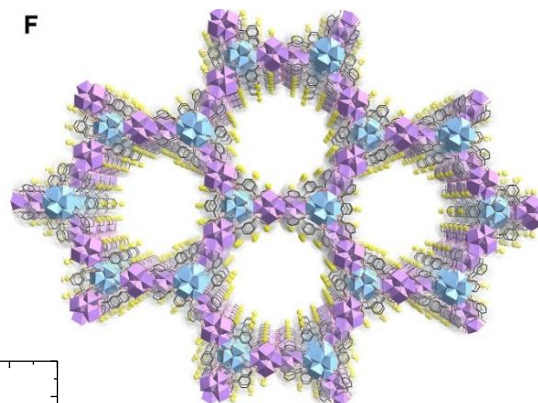
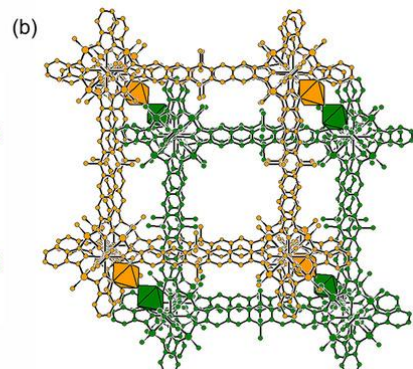
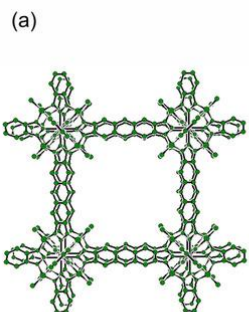


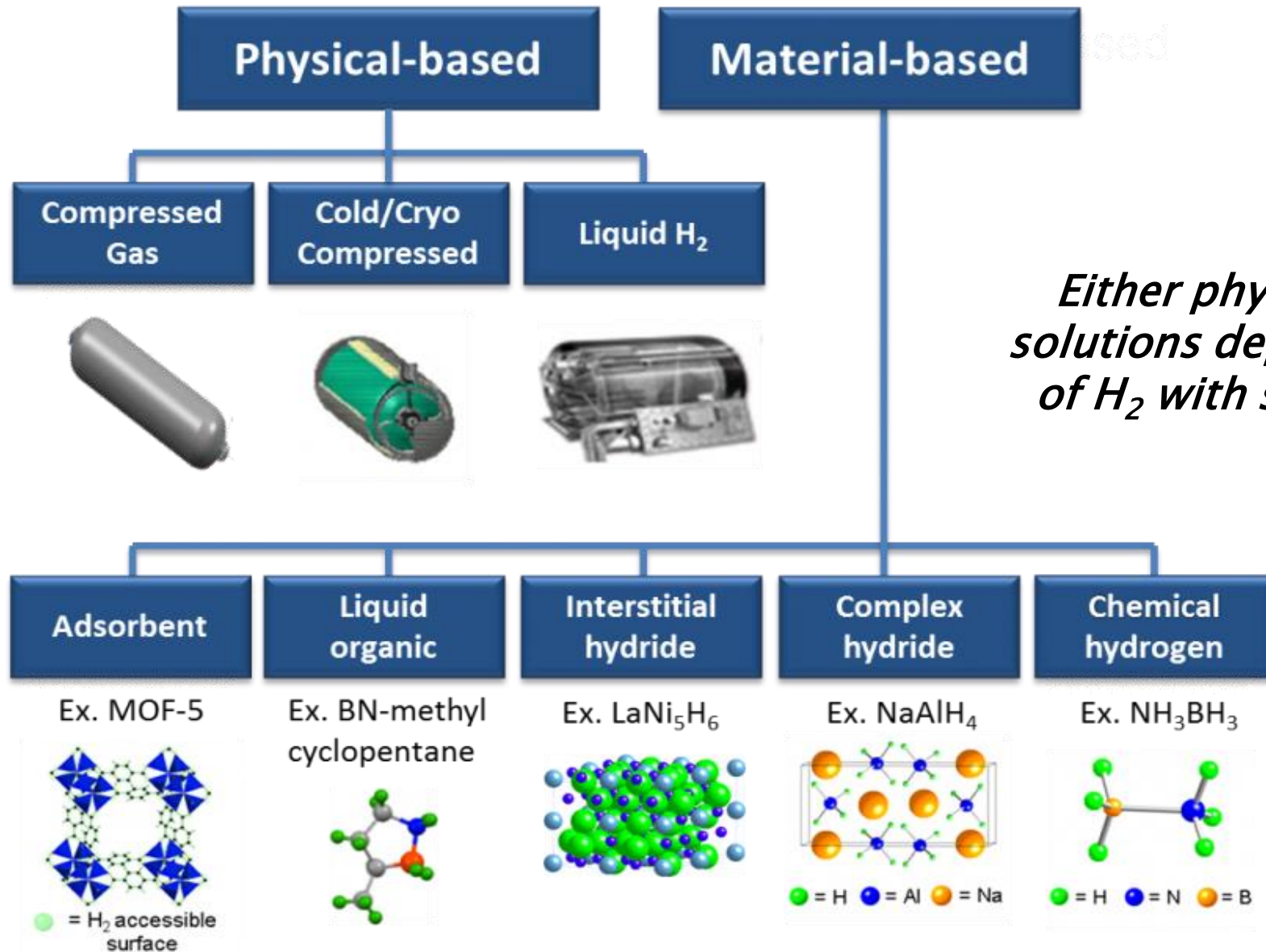
Hydrogen : from storage to production



Christian SERRE – Rencontres CNC – 7 décembre 2023

Institute des Matériaux Poreux de Paris, ENS, ESPCI Paris, université PSL, CNRS, France

Physical vs. material-based solutions for hydrogen storage



Either physical or materials-based solutions depending on the interaction of H₂ with surrounding environment

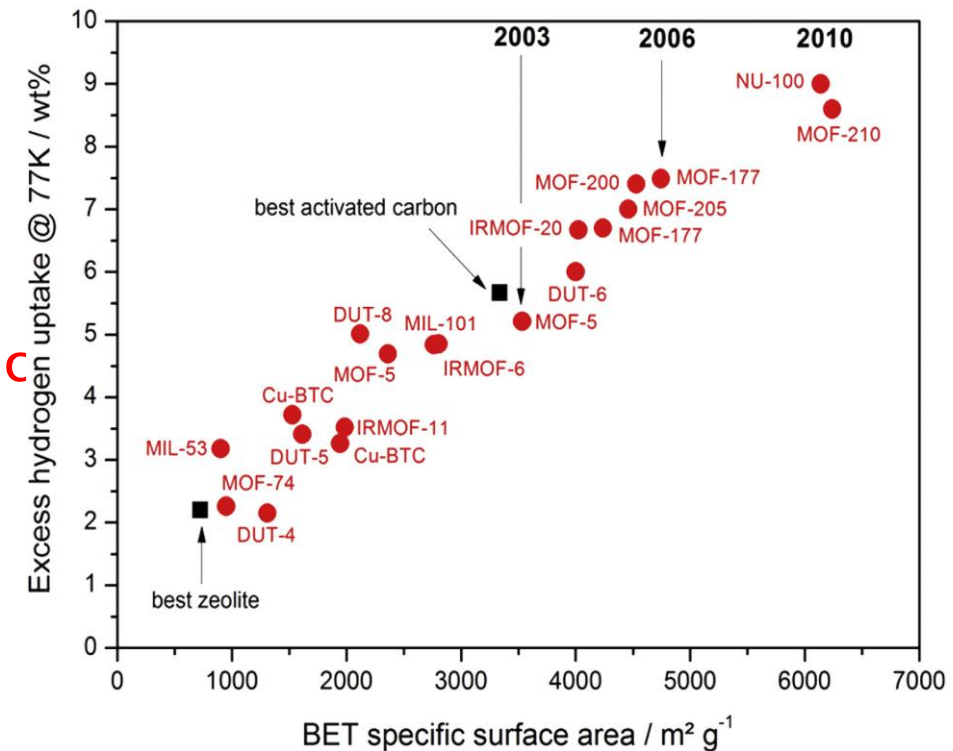
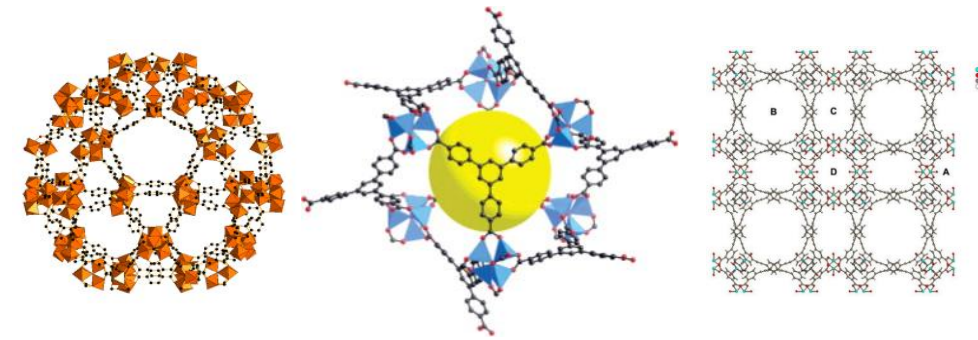
Hydrogen storage in MOFs

- **Advantages**

- High porous volume and specific surface
- Highly tunable chemistry and porosity
- High adsorption capacities at 77 K

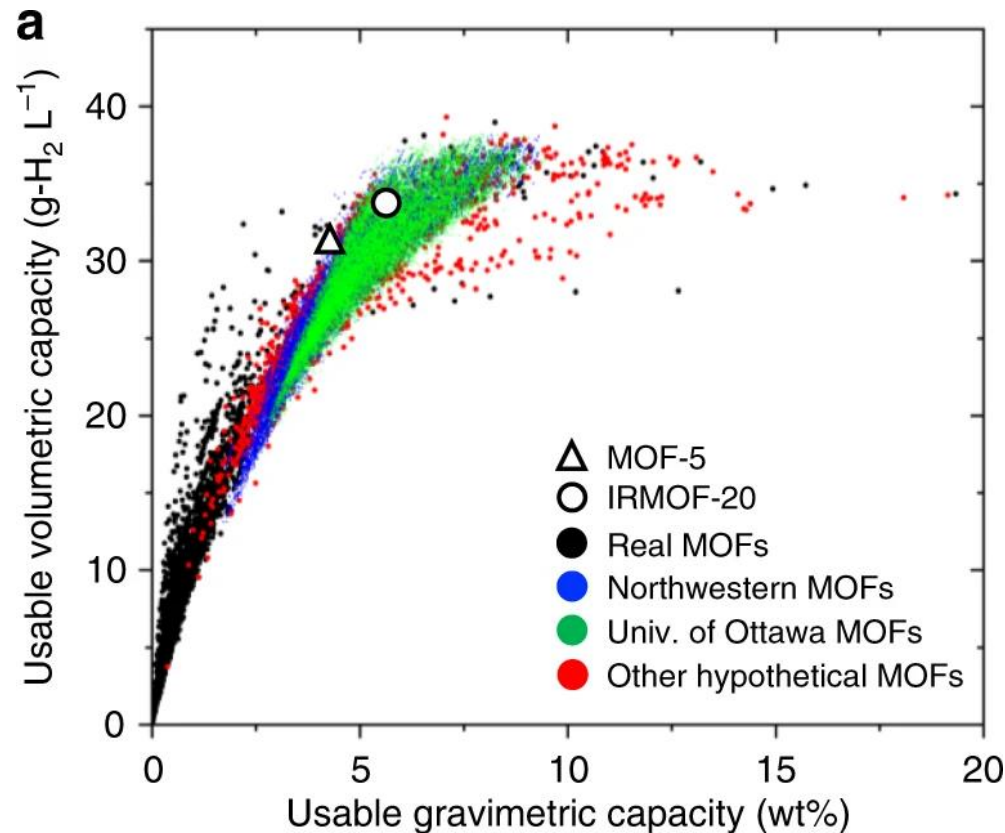
- **Disadvantages**

- Limited volumetric capacity (small density 0,1–1 g/c
- Weak capacities at RT \Leftrightarrow low heat of adsorption
- Limited number of scalable, cheap materials

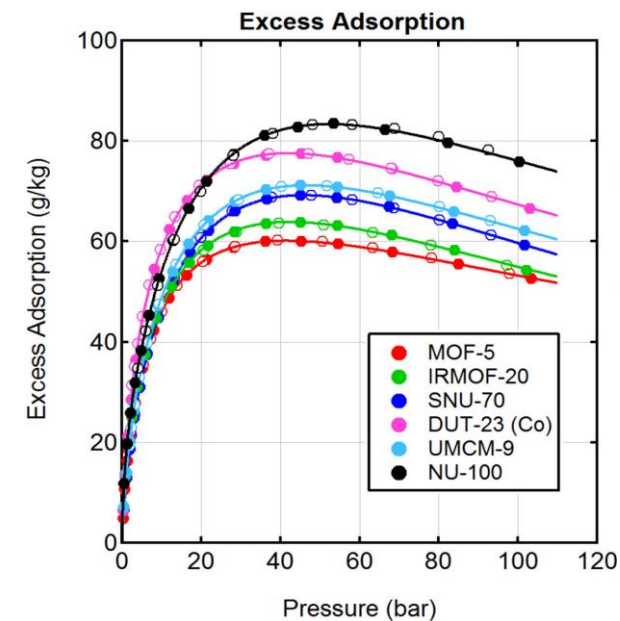


High-throughput computational screening of MOFs – H₂

MOFs : crystalline solids
>25,000 exp. structures
↔ Databases of experimental
And theoretical structures
(500,000)
↔ Datamining, Machine Learning



Usable capacity = accessible uptake upon the pressure swing:
100 bar ↔ 2 bar at 77K

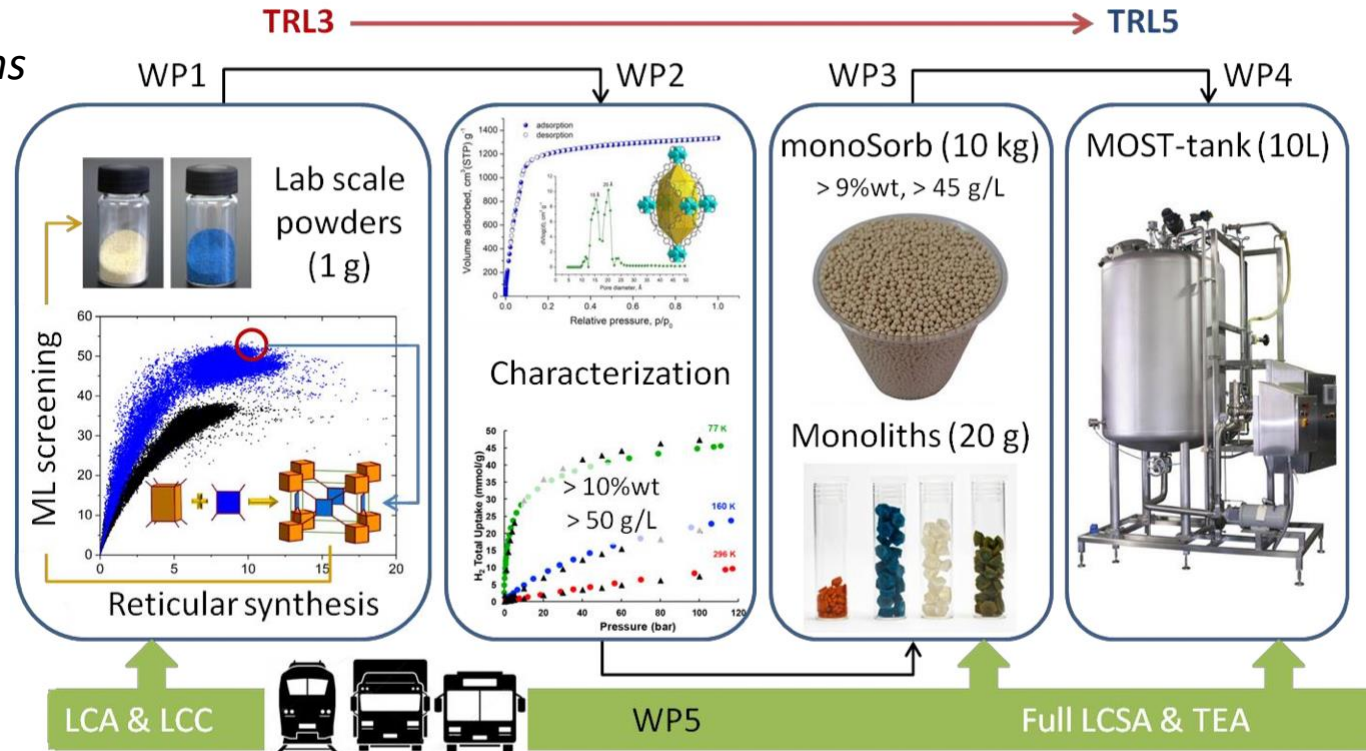


Farha et al. Nature Chem.
2012

- Volumetric capacities increase with gravimetric capacity up to ~10 wt.%, with volumetric performance plateau ~40 g-H₂ L⁻¹
- Capacity can be maximized by increasing gravimetric surface area and porosity

Optimising H₂ storage in MOFs : EU project MOST-H2

Dr Karim Adil
University Le Mans



Horizon-CL4-2021-Resilience-01-07:
Advanced Materials for hydrogen
storage

Multiscale lab-to-tank :
Develop, validate and demonstrate
innovative, low cost, cryo-
adsorptive hydrogen storage

*Monolithic MOFs with optimal
volumetric and gravimetric capacity
VTSA (80-160 K, 5-100 bar)*

Optimal MOF performance & properties targets		monoSorb performance targets	
Usable H ₂ capacity	>10 (wt%), >50(g/L)	Usable H ₂ capacity	>9 (wt%), >45(g/L)
BET Area	>5000 m ² /g, >2500 m ² /cm ³	monoSorb production targets (TRL 5)	
Crystal density	d _c >0.4 g/cm ³	>10 kg	
Porosity	ε >75%	MOST-tank target (TRL 5)	
Pore size	1.5 nm < w < 2.5 nm	500 g H ₂ , ~10 L, ~8-10 kg of monoSorb,	
Total Pore Volume	TPV >2 cm ³ /g	operating between 100 bar@80K and 5 bar@160K	

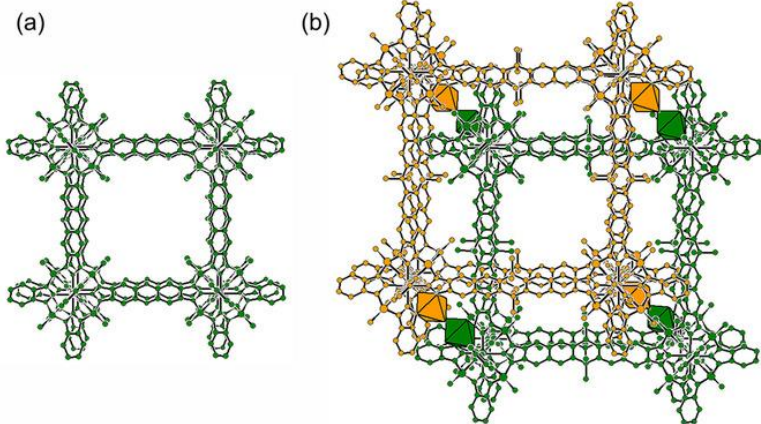
Strategies for improving H₂ storage in MOFs

Interpenetration

Density doubled

Higher stability

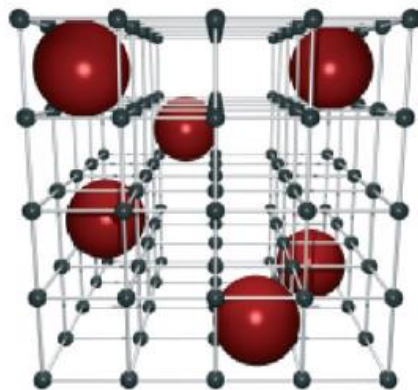
Heat of adsorption increased



Active species loading (Pd⁰, hydrides...)

Heat of adsorption strongly increased

Spillover effect

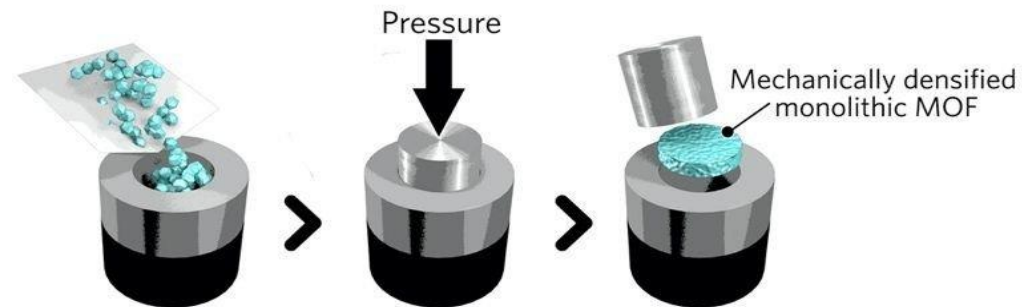


Densification

Density doubled

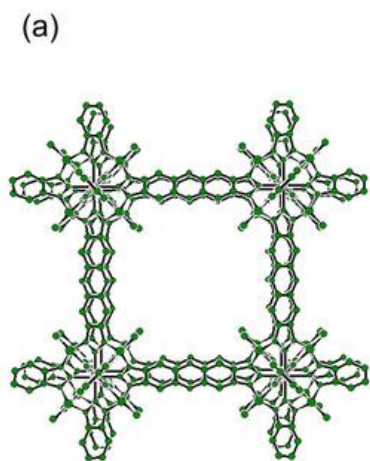
Higher stability

Heat of adsorption increased

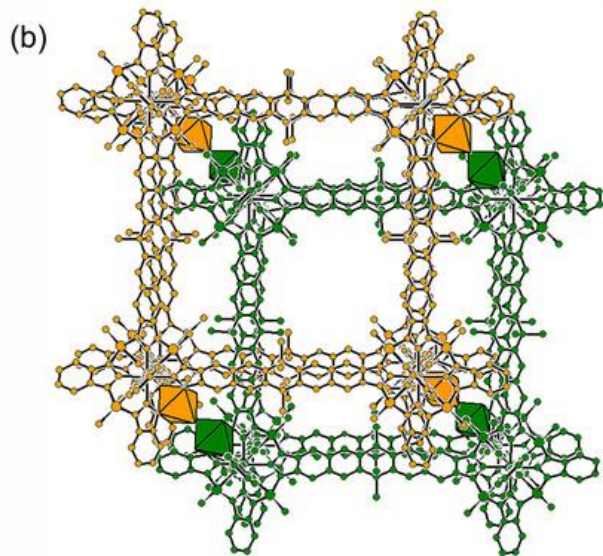


Interpenetration in MOFs for H₂ storage

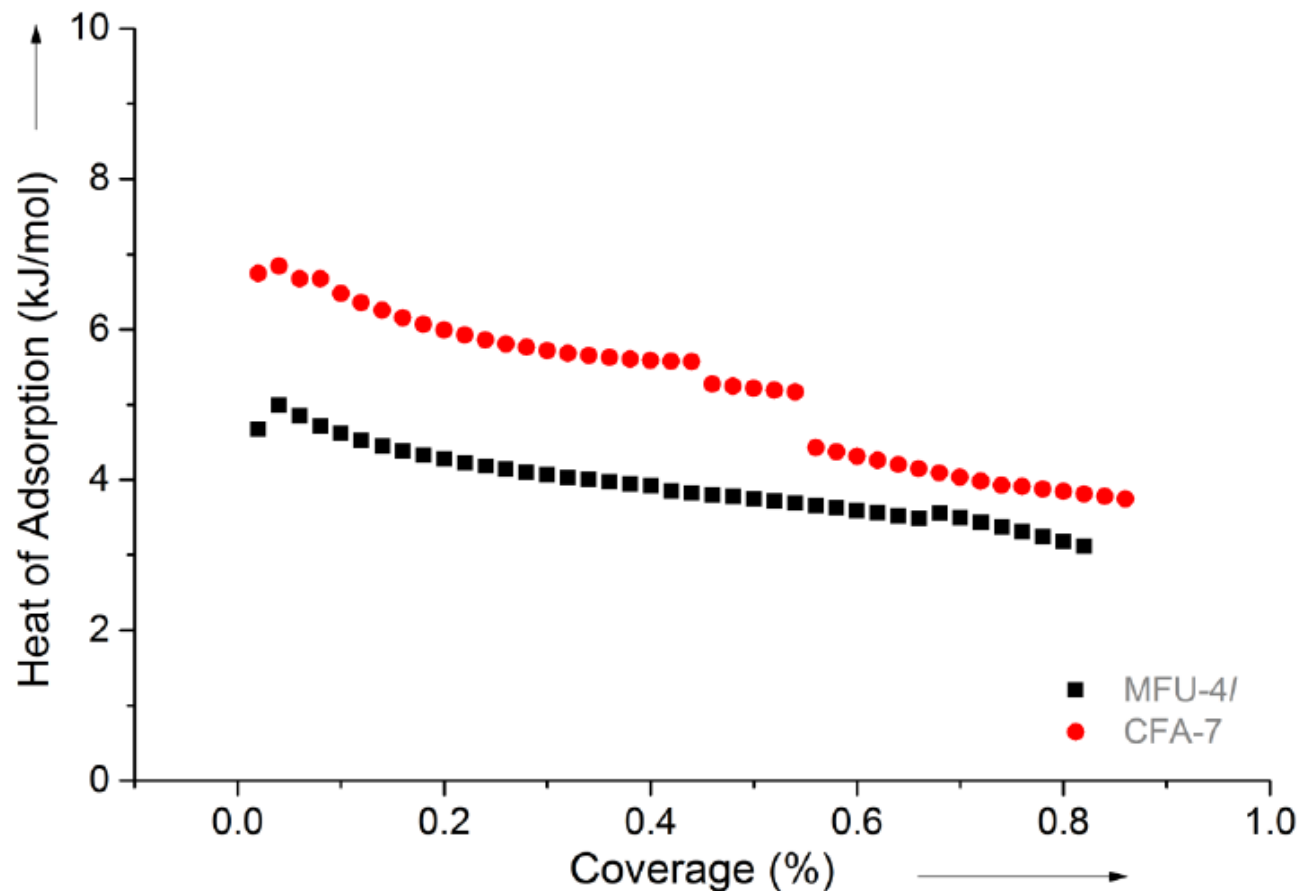
Increasing the H₂ volumetric capacity through interpenetration



MFU-4l
~ 3000 m²/g
 $\rho = 0.5592$ g/mL



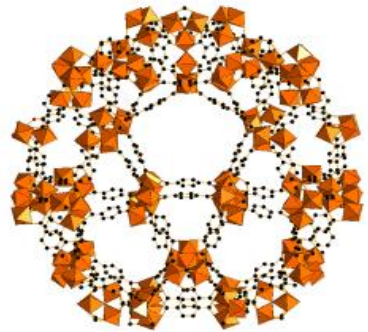
CFA-7
~ 1700 m²/g
 $\rho = 1.570$ g/mL



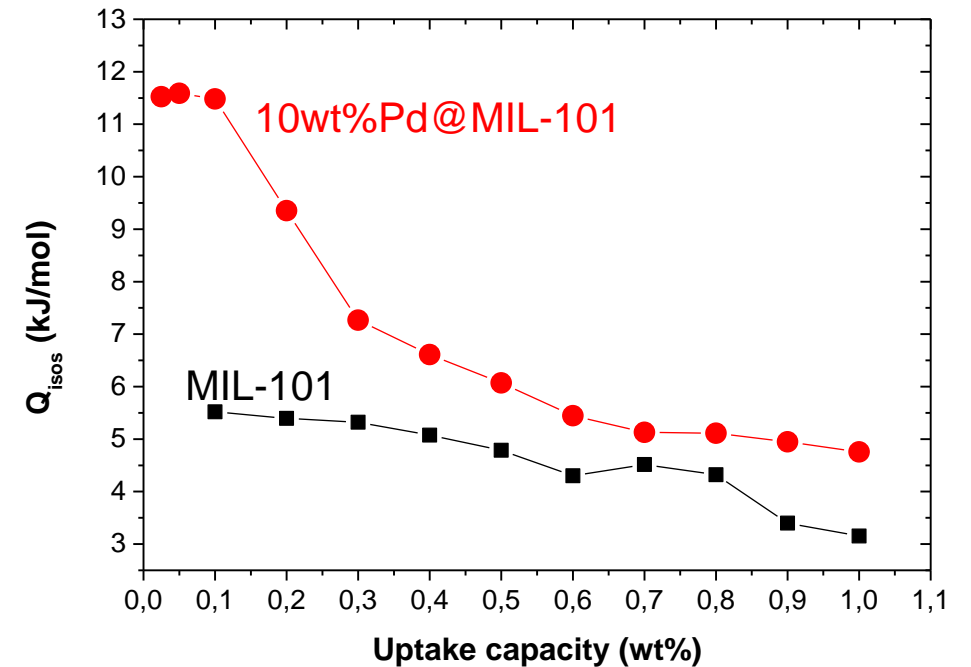
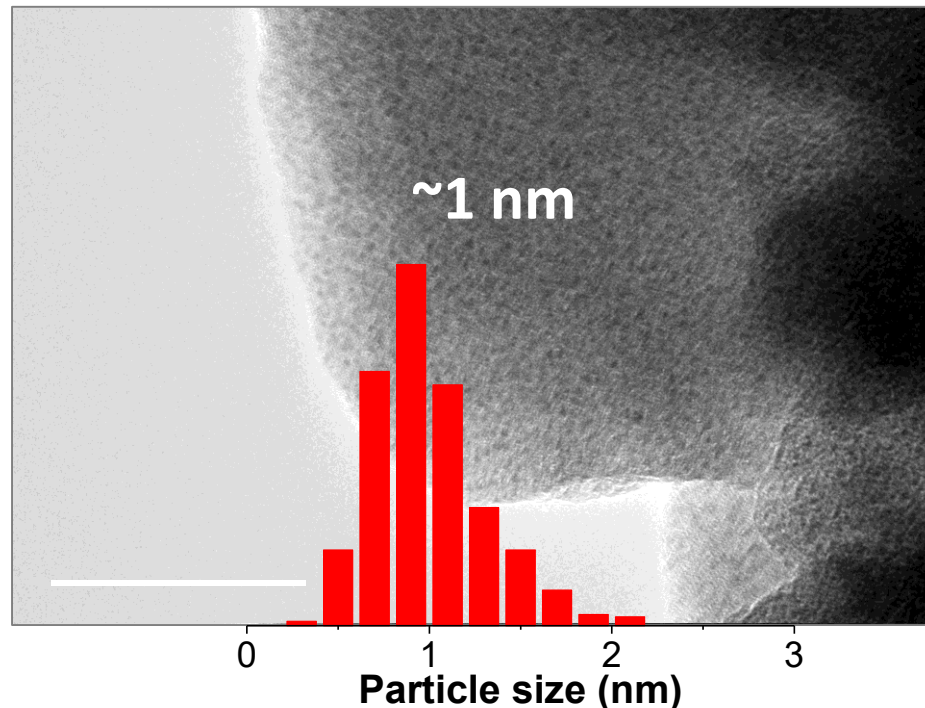
Loading active species to boost the H₂ storage

Dr Claudia Zlotea
ICMPE, CNRS

Increasing the H₂ volumetric capacity : **functionalization** of
MOFs with metal nanoparticles (or metal hydrides)



Mesoporous MOFs
(MIL-101(Cr))



But at the expenses of gravimetric capacity at 77 K

A. Malouche et al., J. Mat. Chem. A, 2017, 5, 23043

C. Roesler and R. A. Fischer, Crystengcomm 2015, 17, 199

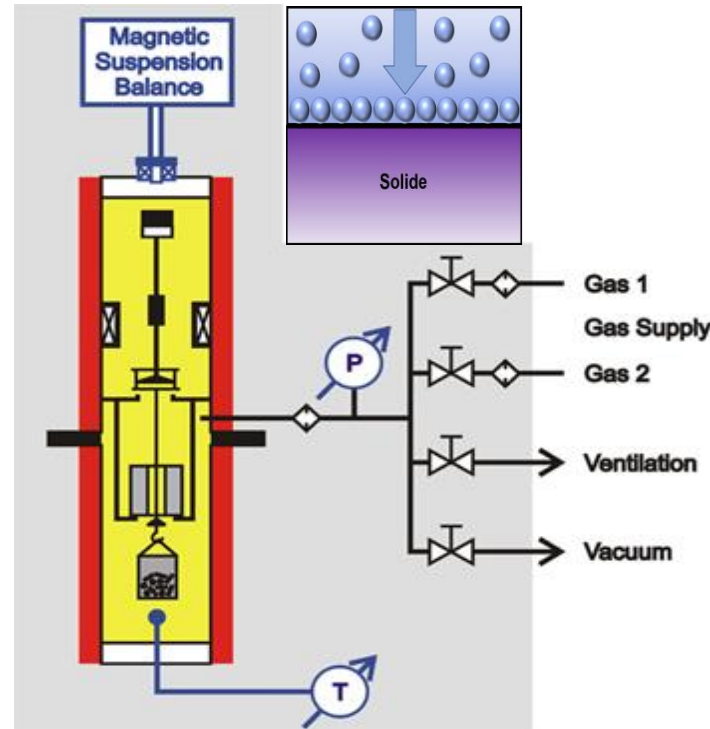
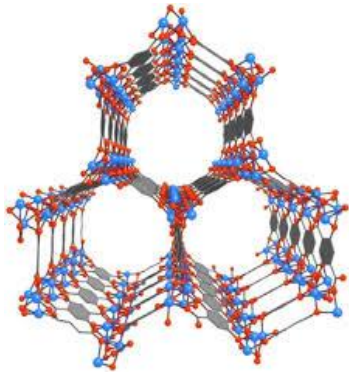
Hydrogen Storage in Metal Doped MOF Composites

ANR Project "PhotoStor2", Coordinator: Prof. Johnny Deschamps (UCP, ENSTA IP Paris),

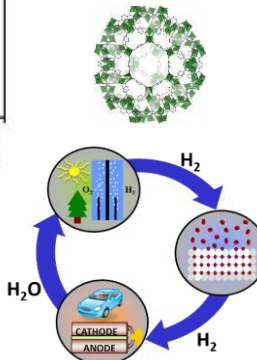
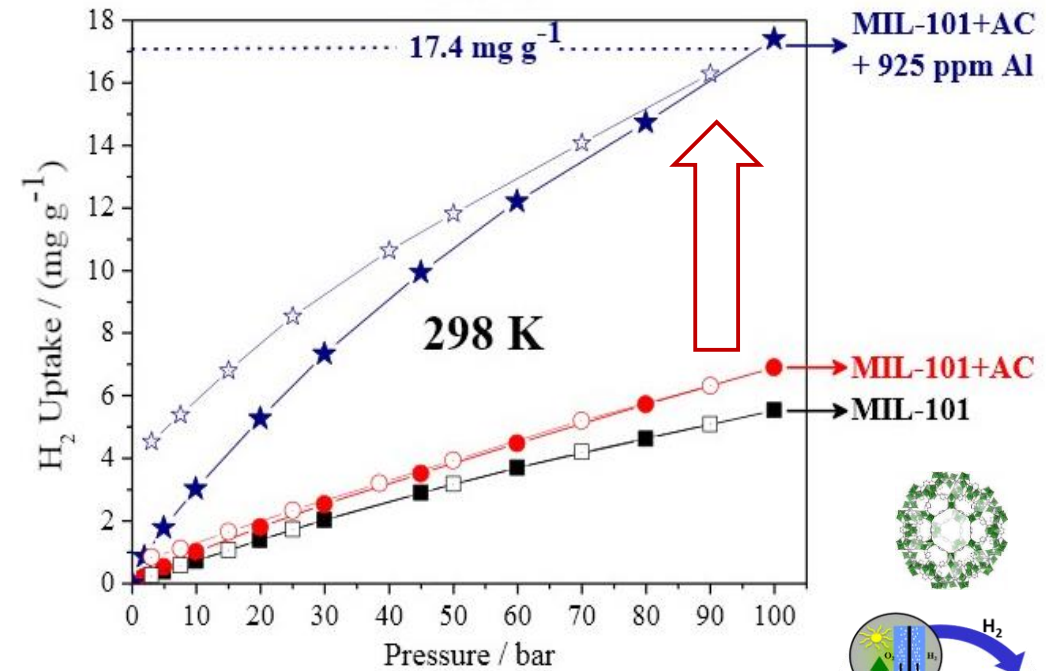
Partner: Dr Hynd Remita (ICP, Paris Saclay)

MOF Materials + Activated Coal (AC) + Metal or Bi-Metallic Nanoparticles

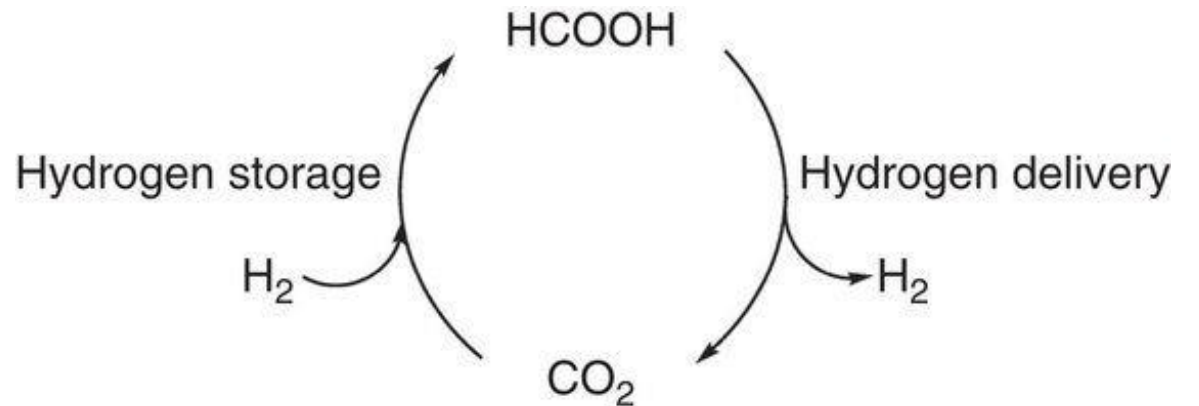
Hydro/solvo-thermal syntheses; Radiolysis; Doping by impregnation in solution or under pressure



*H₂ adsorption at RT
Moderate pressure (max. 100 bar)*



H₂ production from formic acid (1)



Formic acid : short cycle, highly promising for H₂ storage/transportation

Current catalysts : NPs of Palladium in mesoporous solids (water, 60°C, formic acid / formates)

BUT desactivation, degradation thus limiting the practical use

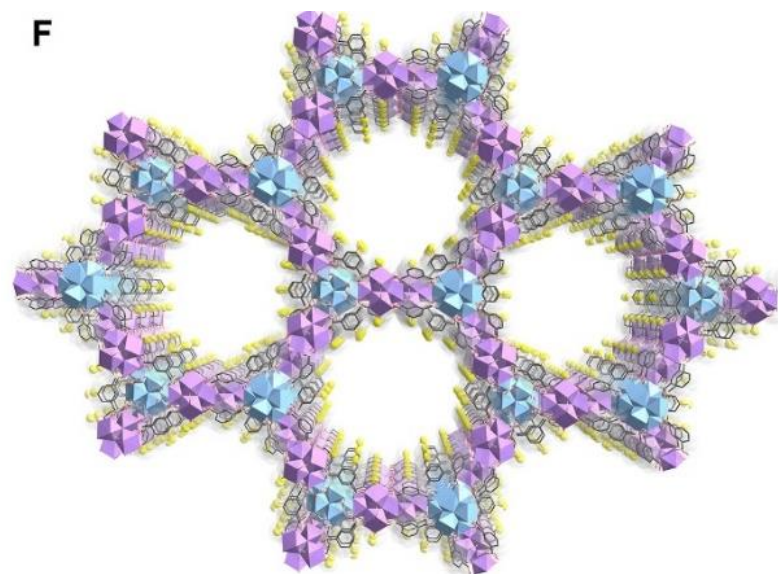
H₂ production from formic acid (2)

Mesoporous Zr-MOFs loaded with Pd NPs (NaBH₄) for production of H₂ from formic acid



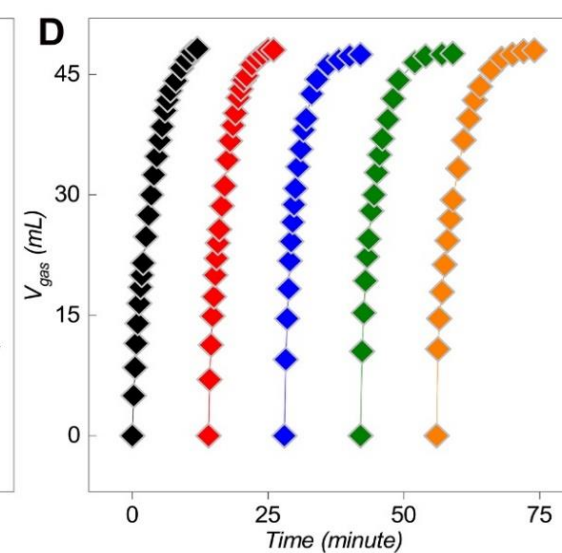
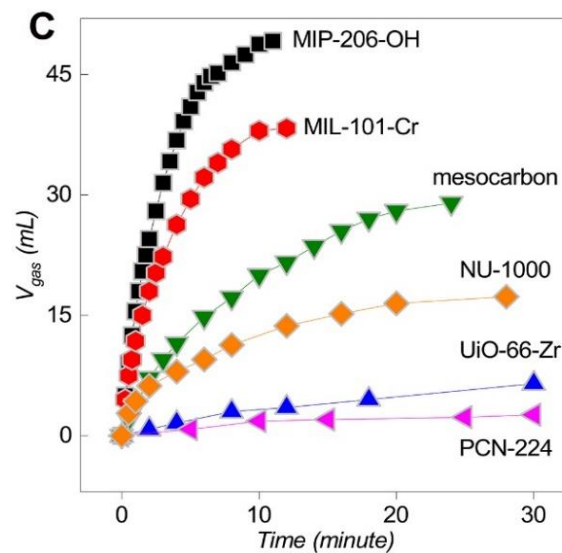
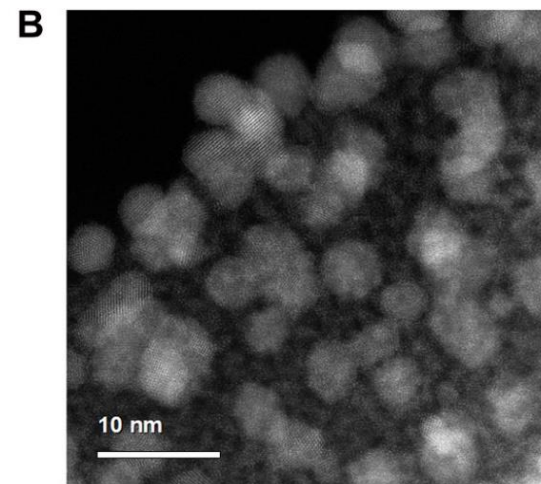
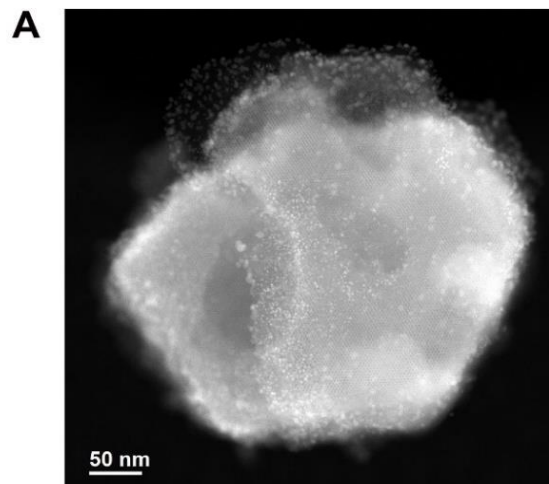
Sujing Wang

Collaboration
Qiang Xu / Liyu Chen
(Osaka /
Shenzen)



*MIP-206 : Zr-1,3 BDC
Mesoporous (2.6 nm)
Highly stable, tunable*

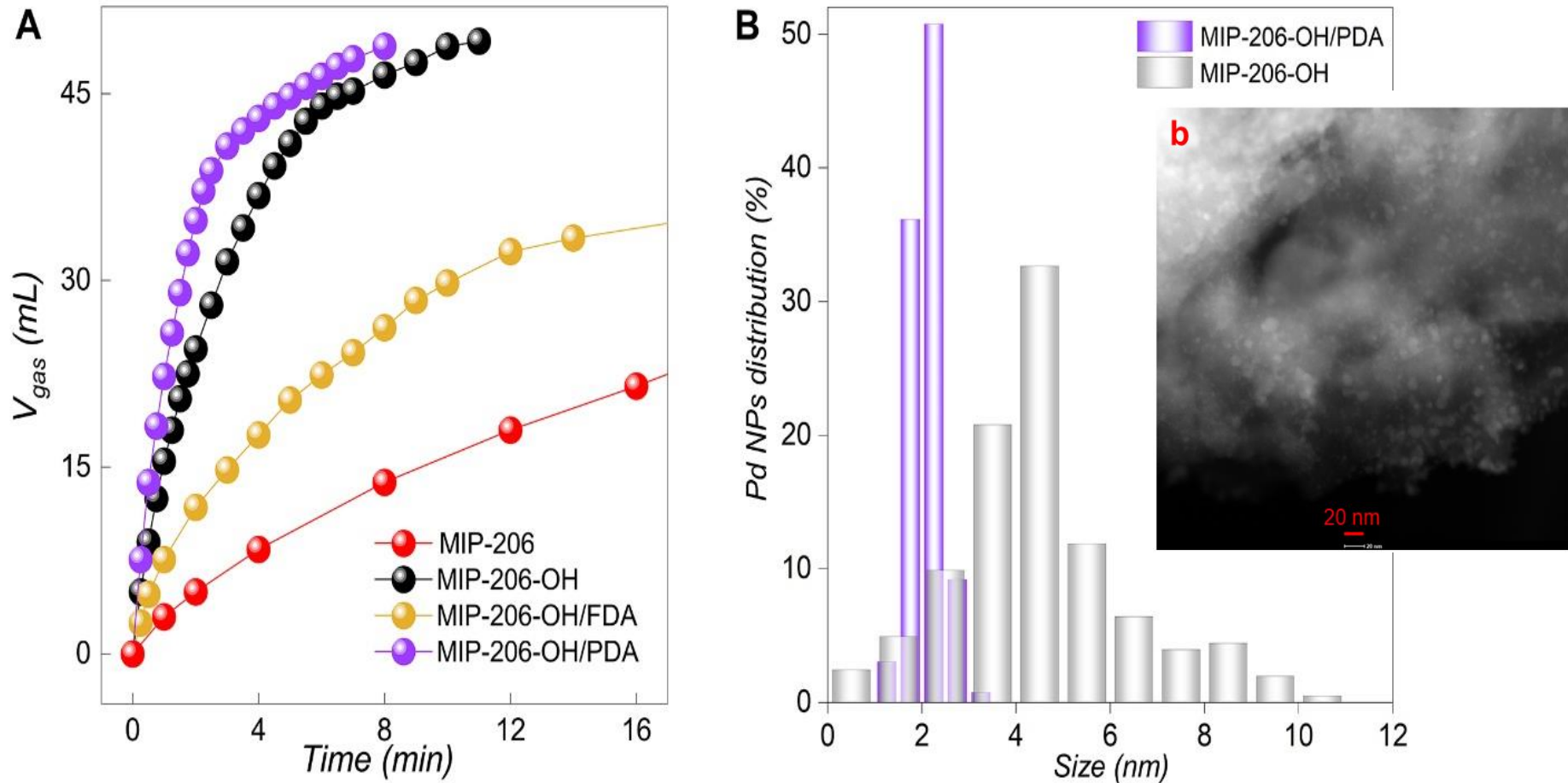
**Among the best composite for the
chemical reduction of formic acid !
Reusable !**



*Water, 60° C
Formic acid / Formates*

H₂ production from formic acid (3)

Control of the Pd NP formation through ligand mixing (MTV)

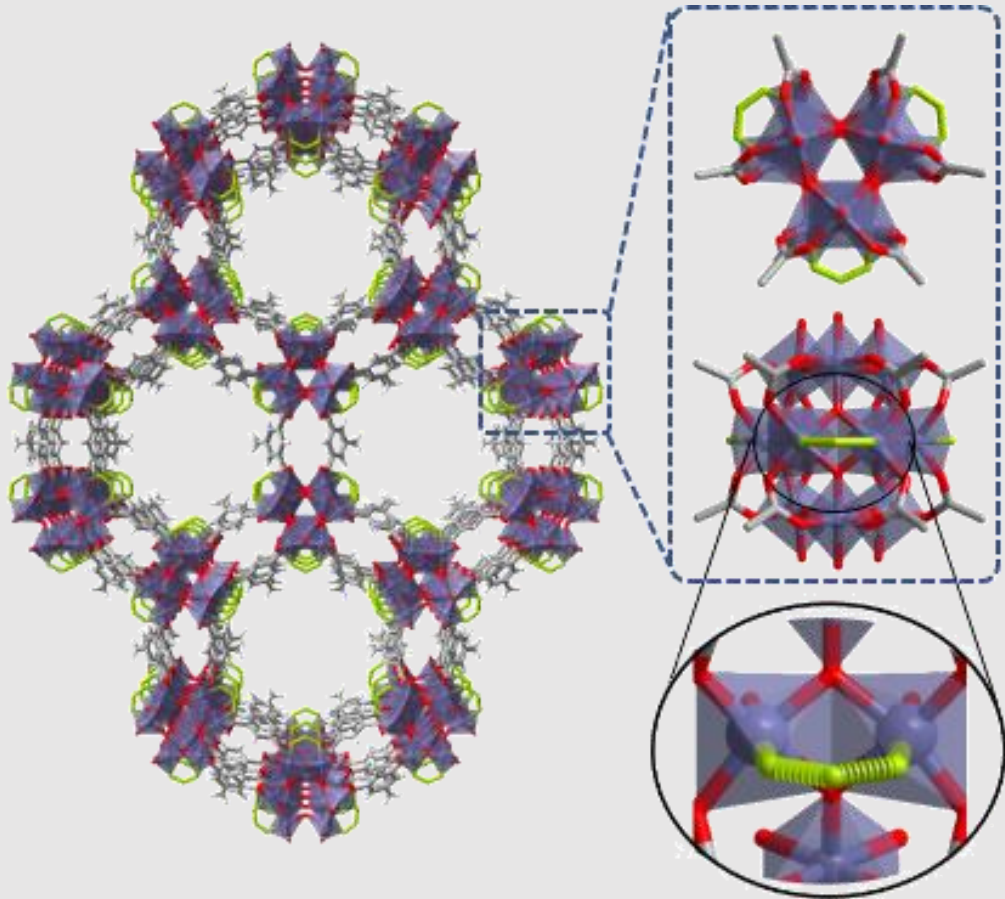


Pd@MTV-MIP-206-OH/PDA (33 % substituted PDA):
→ Smaller NP, lower PSD → inside the pores → higher conversion achieved !!

Photocatalytic H₂ production from formic acid (1)

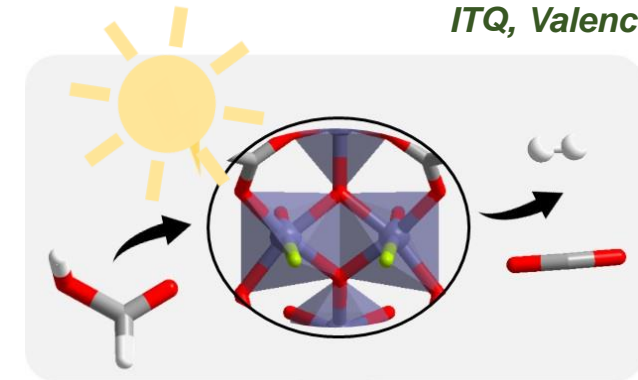
Sergio Navalon & Hermegildo Garcia
ITQ, Valencia, Spain

Ti₁₂O₁₅ oxocluster + accessible metal sites + porosity



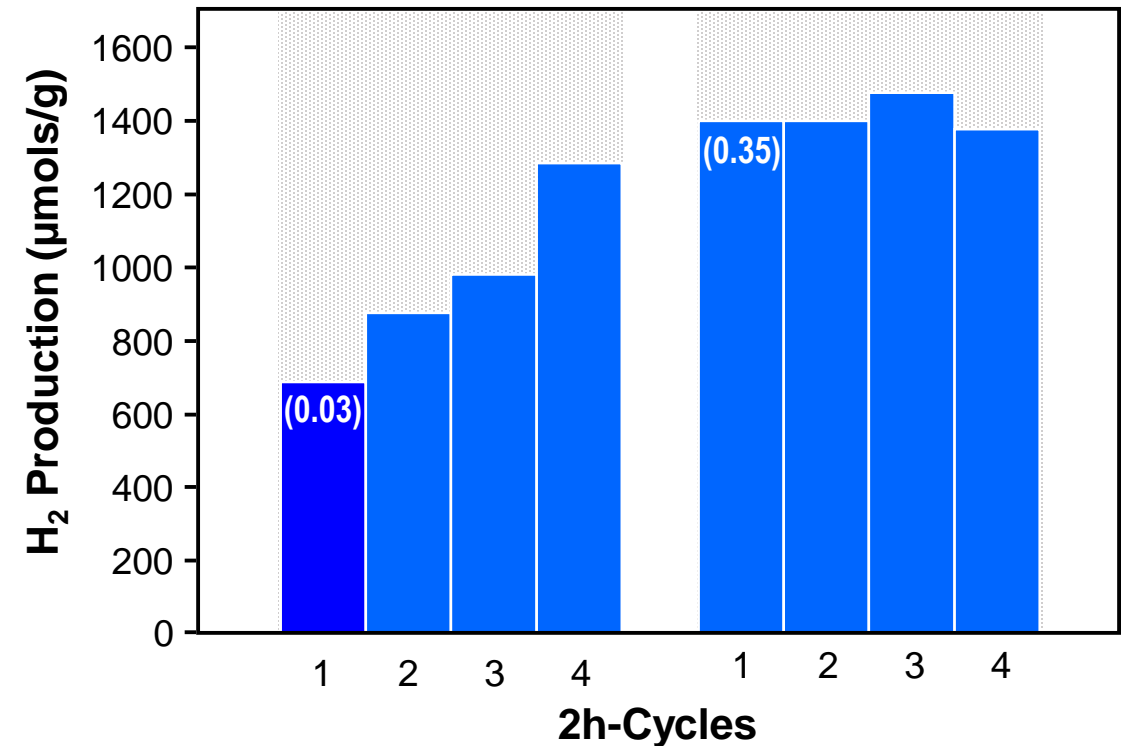
S. Wang *et al.*, *Nat. Commun.* **2018**, 1660

H. García-Baldoví *et al.*, *Energy Environ. Sci.* **2023**, 167



MIP-177_LT

MIP-177_LT-AT



Photocatalytic H₂ production from formic acid (2)

Sergio Navalon & Hermegildo Garcia
ITQ, Valencia, Spain



INSTITUTO DE
TECNOLOGÍA
QUÍMICA

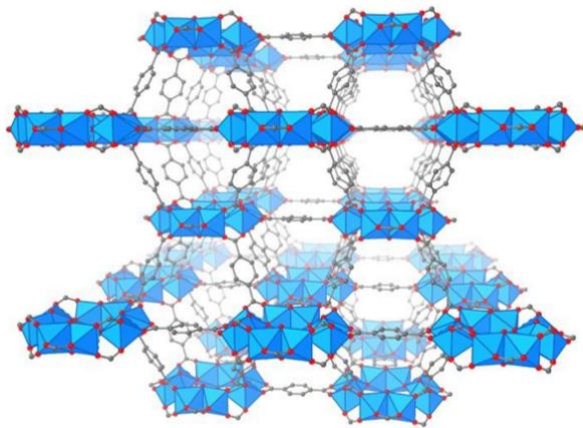
CSIC
CONSEJO SUPERIOR DE INVESTIGACIONES CIENTÍFICAS

UNIVERSITAT
POLITÀCNICA
DE VALÈNCIA

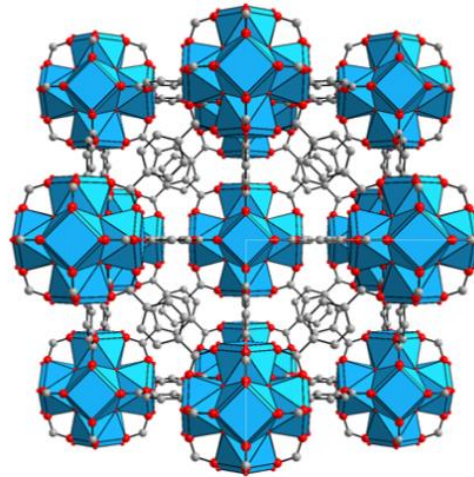
MIP-177_LT outperforms other MOFs and TiO₂

300 W arc Xe lamp
10 mg in 20 mL of 10⁻³ M FA/H₂O

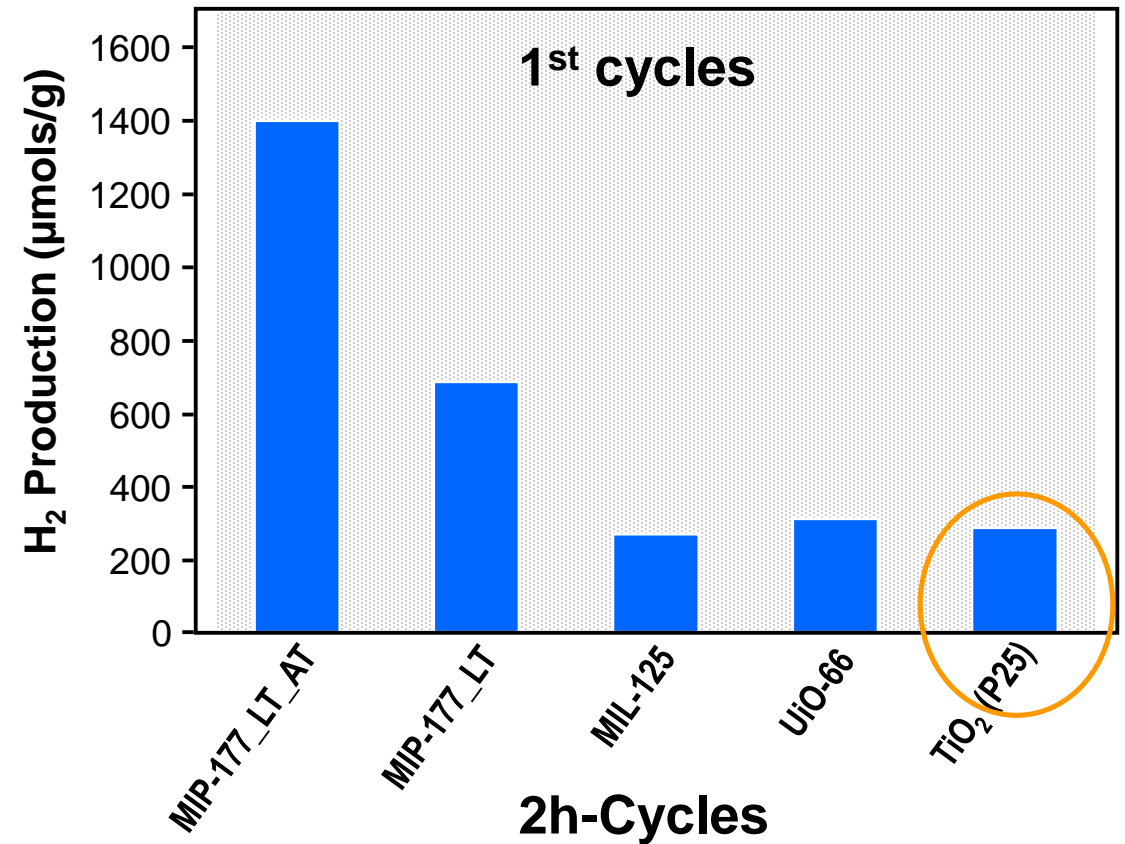
Benchmarking



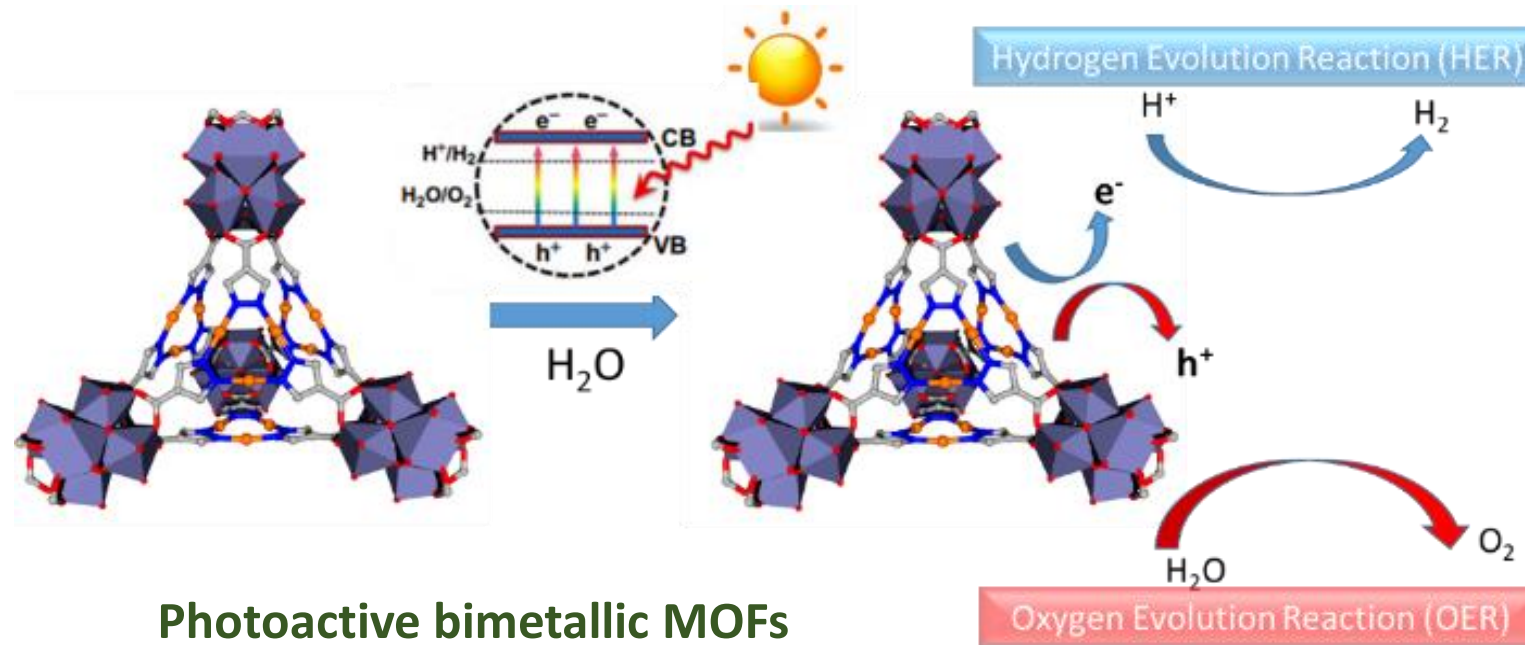
MIL-125



UiO-66



Photocatalytic H₂ production from Overall Water Splitting



Photoactive bimetallic MOFs

MOF2H2 : Horizon-CL5-2021-D3-03:
Sustainable, secure and competitive
energy supply

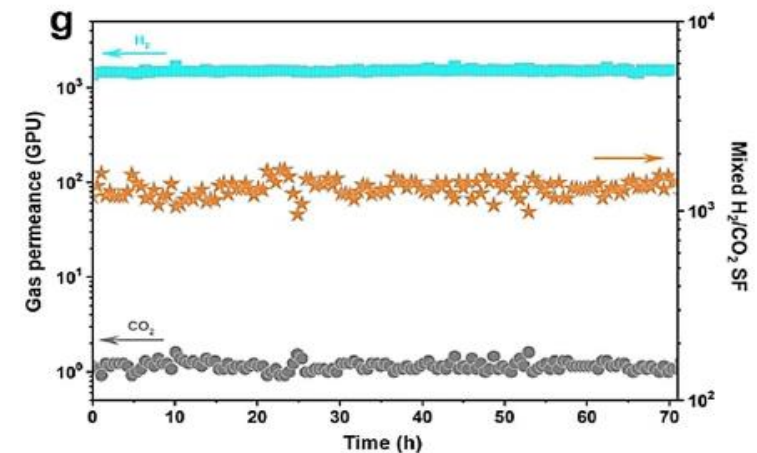
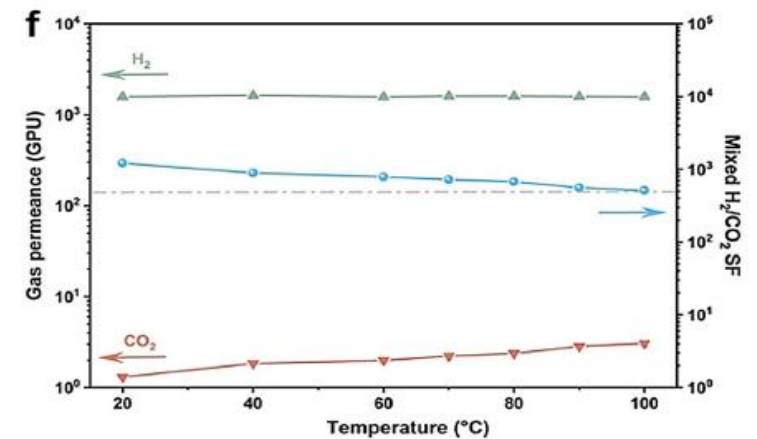
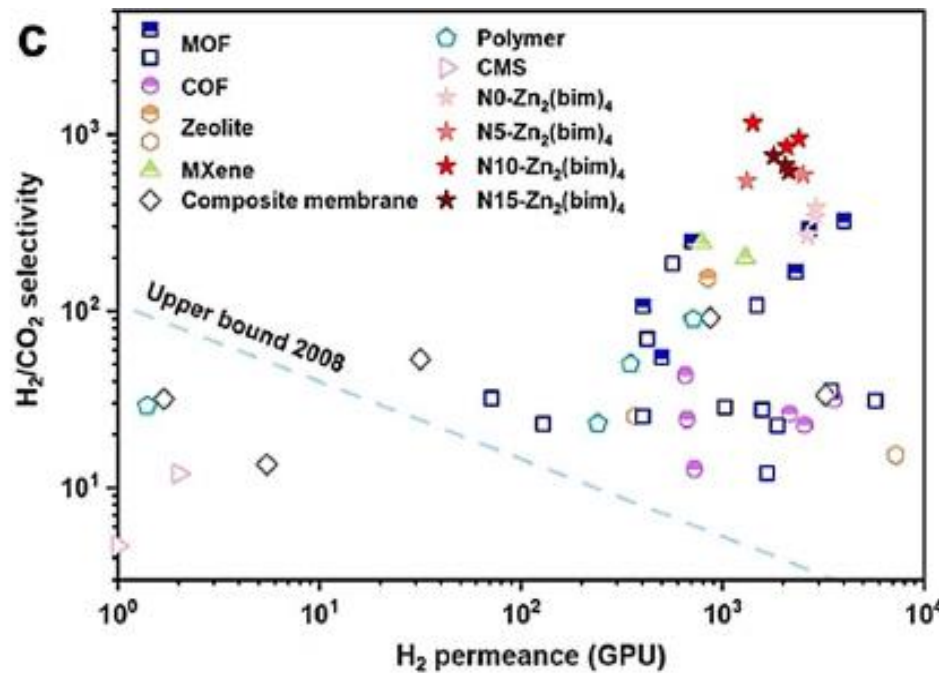
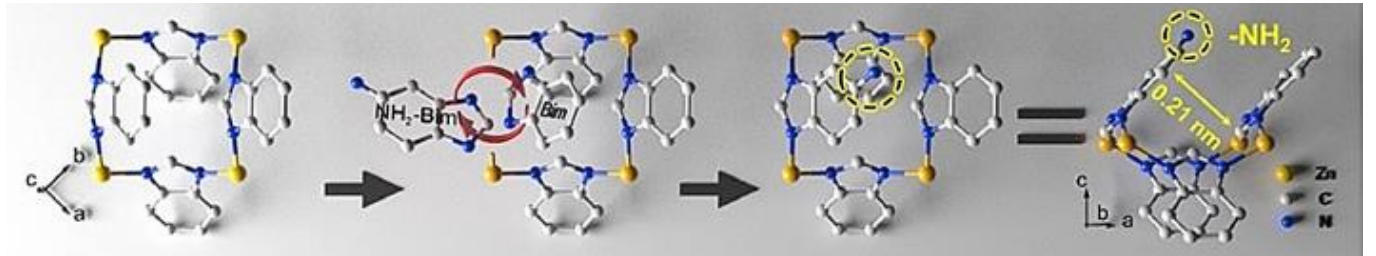


Coordinator:
ESPCI – Paris
(C. Serre)

- Noble metal free MOFs record-breaking STH for Overall Water Splitting under visible light, with an STH of >2-5%
- Laboratory-scale photocatalytic system, based on an easily green, scalable MOF to efficiently perform the OWS using sunlight irradiation and with long-term stability

MOFs for the purification of hydrogen

Microporous Zn benzimidazole
Functionalised (amino groups) membrane

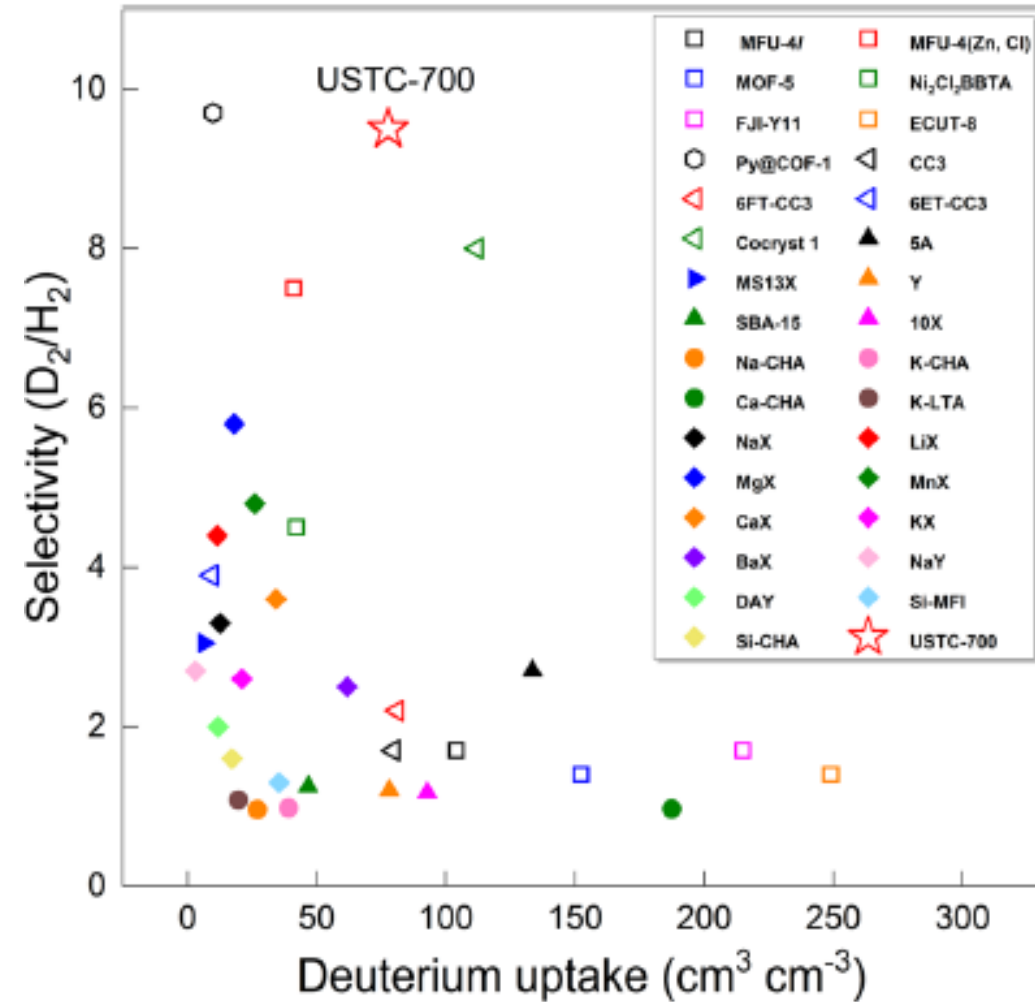
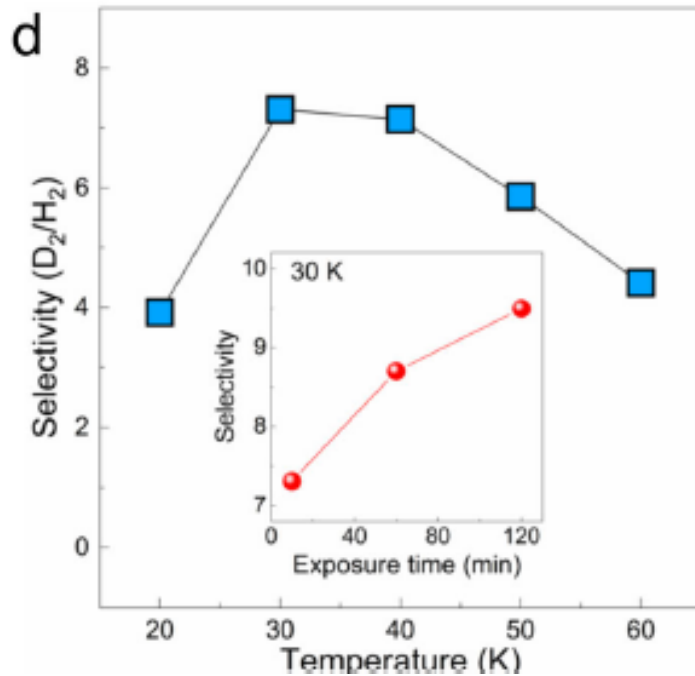
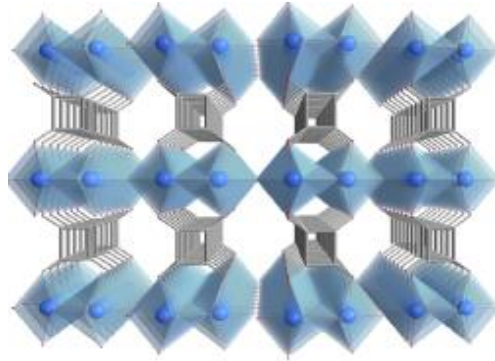


Exceptional H_2/CO_2 selectivity and permeability

W. Yang et al, Angew. Chem. Int. Ed. 2023

MOFs for the H₂/D₂ (cryogenic) separation

Microporous Ti squarate



Exceptional H₂/CO₂ selectivity and capacity

G. Maurin, S. Yang et al, Nat. Comm. 2023

Acknowledgments

- ❑ Post-docs/Students : Sujing Wang, Raquel Del Angel Montes
- ❑ Colleagues : George Mouchaham, Antoine Tissot, Iurii Dovgaliuk

- ❑ Maria Cabrero Antonino, Sergio Navalone, Hermenegildo Garcia (*Valence, Espagne*)

- ❑ Liyu Chen, Qiang Xu (*Osaka, Japon / Shenzen, Chine*)

- ❑ Mohammad Wahiduzzaman, Dong Fan, Guillaume Maurin (*Montpellier, France*)

- ❑ Eric Elkaim (*Cristal*), William Shepard (*PX2*) (*Soleil, France*)

- ❑ Claudia Zlotea (*ICMPE, Thiais, France*)

- ❑ Karim Adil (*Le Mans, France*)

- ❑ Johnny Deschamps, (*Ensta, France*)

- ❑ Hynd Remita (*ICP, Saclay, France*)



